

## Chapter 7 Ionic And Metallic Bonding Vocabulary Review Answers

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Although they are composed of ions, ionic compounds are electrically neutral. the electrostatic forces that hold ions together in ionic compounds. shows the kinds and numbers of atoms in the smallest representative unit of a substance. the lowest whole-number ratio of ions in an ionic compound.

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Chapter 7 - Ionic and Metallic Bonding Section 7.3 - Bonding in Metals. The valence electrons of metal atoms can be modeled as a sea of electrons. Metallic bonds consist of the attraction of the free-floating valence electrons for the positively charged metal ions.

### Chapter 7 Ionic And Metallic Bonding Test Answers

ionic compound. they are electrically neutral and composed of cations and anions, generally have high melting points and are crystalline at room temperature, and they can conduct an electric current when melted or dissolved in water. ionic bonds. the electrostatic forces that hold ions together in ionic compounds.

### Pearson Chemistry Chapter 7 "Ionic and Metallic Bonding ...

Chapter 7 - Ionic and Metallic Bonding. Jennie L. Borders. Section 7.1 - Ions. Valence electrons are the electrons in the highest occupied energy level. Valence electrons are the only electrons involved in chemical bonding. Elements in the same group have the same number of valence electrons.

## Chapter 7 - Ionic and Metallic Bonding

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Chapter 7: Ionic Bonding (Sections 9.1, 9.2) Section 7.1 - Ions •Valence electrons: electrons in the highest occupied energy level. •The only electrons involved in chemical bonding. •Elements in the same group have the same number of valence electrons. Lewis Dot Diagrams Show the symbol of the element surrounded by the valence electrons ...

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Chapter 7 - Ionic and Metallic Bonding - 7.1 Ions - 7.1 Lesson Check - Page 199: 1. Answer. To find the number of valence electrons in an a representative element simply look at the group number.

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## CHAPTER 7 IONIC AND METALLIC BONDING ANSWER KEY PEARSON ...

►►Elements in Group 1 form cations with a charge of 1+, and those in Group 2 form cations with a charge of 2+. ►►Many transition metals form more than one cation and do not follow the octet rule. Formation of Anions Anions are negatively charged ions formed when an atom gains one or more valence electrons.

## BONDING AND INTERACTIONS

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## Ionic And Metallic Bonding Chapter 7 Worksheet Answers

7. Define Alloy. A compound of cations and anions; An anion of a halogen; A mixture of two or more elements, at least one of which is a metal; Lowest whole number ratio of ions in an ionic compound; Valence electrons and positively charged metal ions

## Chemistry Chapter 7-Ionic and Metallic Bonding

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