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4. Is the indicated compound acting an acid or a base in the following reaction? $\text{O}=\text{O} + \text{H}_2\text{SO}_4$
 $\text{HSO}_4^- + \text{O}=\text{O} \rightleftharpoons \text{O}=\text{O} + \text{HSO}_4^-$
A. Acid B. Base C. Neither Ans: B Topic: Introduction to Bronsted-Lowry Acids and Bases

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